

1. If the activity of a substance drops by a factor of 32 in 5 seconds, what is the radioactive half-life?
  
2. Can a given nucleus have both  $\beta^+$  and  $\beta^-$  decay modes? If yes, Give an example.
  
3. The  $\alpha$ -decay of a  $^{238}\text{Pu}$  ( $\tau = 127$  years) nuclide into a long lived  $^{234}\text{U}$  ( $\tau = 3.5 \cdot 10^5$  years) daughter nucleus releases 5.49 MeV kinetic energy. The heat so produced can be converted into useful electricity by radio-thermal generators (RTG's). The *Voyager 2* space probe, which was launched on 20/8/1977, flew past four planets, including Saturn, which it reached on 26/8/1981. Saturn's separation from the sun is 9.8 AU.
  - (a) How much plutonium would an RTG on Voyager 2 with 5.5% efficiency have to carry so as to deliver at least 395 W electric power when the probe flies past Saturn?
  - (b) How much electric power would then be available at Neptune (24/8/1989, 30.1 AU)?
  - (c) Where is Voyager 2 today?
  - (d) How much power delivers the RTG?
  
4. Naturally occurring uranium is a mixture of the  $^{238}\text{U}$  (99.28%) and  $^{235}\text{U}$  (0.72%) isotopes.
  - (a) How old must the material of the solar system be if one assumes that at its creation both isotopes were present in equal quantities? ( $^{235}\text{U}$ :  $\tau = 1.015 \cdot 10^9$  years.  $^{238}\text{U}$ :  $t_{1/2} = 4.5 \cdot 10^9$  years)
  - (b) How much energy per uranium nucleus is set free in the decay chain  $^{238}\text{U} \rightarrow ^{206}\text{Pb}$ ?